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AS Unit F331: Chemistry for Life

Unit 1: Elements of Life (EL)

A study of elements and compounds in the universe, the solar system and the human body.

The chemical ideas in this module are:

- atomic structure;
- radioactivity: fission and fusion;
- chemical equations and amount of substance;
- the Periodic Table and Group 2 chemistry;
- bonding and the shapes of molecules.

Topic	Number	Assessable learning outcomes	Number of times it has been examined	CGP-AS Revision guide (Page number)	Chemical Ideas (Page number)
Formulae, equations and amount of substance	EL1i	Explain and use the term mass number.	1	4-5	16-17
	EL1ii	Explain and use the term relative atomic mass.	3	5, 8, 9, 26	1, 3
	EL1iii	Explain and use the term atomic number.	3	4-5	16-17, 237
	EL1iv	Explain and use the term isotope.	3	5, 8, 9-10	17-23
	EL1v	Explain and use the term Avogadro constant.	1	12	3-4, 10
	EL1vi	Explain and use the term relative isotopic mass.	-	8	17
	EL1vii	Explain and use the term relative formula mass.	-	8	1, 3
	EL1viii	Explain and use the term relative molecular mass.	2	8-12	139
	EL2i	Use the concept of amount of substance to perform calculations involving masses of substances.	2	13-15	1-15
	EL2ii	Use the concept of amount of substance to perform calculations involving empirical and molecular formulae.	2	13-15	1-15
	EL2iii	Use the concept of amount of substance to perform calculations involving percentage composition.	4	13-15	1-15
	EL3	Write and interpret balanced chemical equations including state symbols.	4	12, 52	7-11
	Atomic structure	EL4	Describe protons, neutrons and electrons in terms of their mass and relative charge.	1	4-7

Dr A. Johnston, Southampton, 2014

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